
Perfluorinated Chemicals in Sediments, Lichens, and Seabirds from the Antarctic Peninsula – Environmental Assessment and Management Perspectives

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Abstract

Antarctica is one of the last frontiers of the planet to be investigated for the environmental transport and accumulation of persistent organic pollutants. Perfluorinated contaminants (PFCs) are a group of widely used anthropogenic substances, representing a significant risk to wildlife and humans due to their high biomagnification potential and toxicity risks, especially in food webs of the northern hemisphere and Arctic. Because the assessment of PFCs in the Antarctic continent is scarce, questions linger about the long-range transport and bioaccumulation capacity of PFCs in Antarctic food webs. To better understand the global environmental fate of PFCs, sediment, lichen (*Usnea aurantiaco-atra*), and seabird samples (southern giant petrel, *Macronectes giganteus*; gentoo penguin, *Pygoscelis papua*) were collected around the Antarctic Peninsula in 2009. PFC analytes were analyzed by LC/MS/MS, revealing the detection of PFHpA in seabirds' feather and fecal samples, and PFHxS in lichens. PFBA and PFPeA were detected in 80% and 60% of the lichens, and PFTA in 60% of sediment samples. While oceanic currents and atmospheric transport of PFCs may explain the ubiquitous nature of these contaminants in the Antarctic Peninsula, military bases and research stations established there may also be contributing as secondary sources of PFCs in the Antarctic ecosystem.

Keywords: Perfluorinated compounds, biota, sediment, Antarctic Peninsula, global transport

1. Introduction

Past research shows that legacy persistent organic pollutants (POPs) such as dichlorodiphenyltrichloroethane (DDT), polychlorinated biphenyls (PCBs), and hexachlorocyclohexanes (HCHs) pose substantial problems related to environmental and ecosystem health on a global scale [1–5]. POPs can be transported over very long distances, biomagnify in food webs, and cause adverse health effects in high trophic level species such as birds and marine mammals. Cold regions that are typically isolated from anthropogenic activity, such as the Arctic and the Antarctic, are particularly vulnerable to POPs because of the global distillation phenomenon, which causes many pollutants to concentrate in these regions [6, 7]. The Arctic Monitoring Assessment Program (AMAP), in association with the United Nations Environment Programme (UNEP) Stockholm Convention on Persistent Organic Pollutants, has played a key role in documenting the fate, transport, and effects of these pollutants in the Arctic, and has promoted global initiatives to monitor, manage, and control these substances [6, 8]. Despite enhanced understanding of POP contamination in the Arctic, limited information exists on the state of pollution in Antarctic food webs. Researchers have identified a lack of comparative data between the polar regions of the world, where many efforts have been directed toward understanding POP contamination in the high latitudes of the Northern Hemisphere such as the Canadian Arctic and Greenland [8–12].

Ongoing research has identified emerging contaminants of concern, including perfluorinated contaminants (PFCs), which are expected to pose significant risks to the environment and wildlife, particularly in the Arctic and the Antarctic [13–15]. Although PFCs have been detected in some Antarctic ecosystems and biota, the environmental transport and bioaccumulation patterns of PFCs, mainly perfluoroalkyl acids (PFAAs) such as perfluorinated carboxylates (PFCAs) and perfluorinated sulfonates (PFSAs), remain relatively unexplored within Antarctica. PFCs are highly fluorinated anthropogenic compounds, often utilized as repelling agents, with applications including coatings for paper or food packaging and textiles, industrial surfactants, insecticides, and historically, aqueous film-forming foams [16,17]. Due to their widespread use, PFCs are now considered environmentally ubiquitous substances, found in all areas around the world. In response, numerous measures have been taken to reduce the adverse impacts of PFCs on local and global scales [8].

PFCs are extremely persistent, can travel long distances (predominantly via ocean currents), bioaccumulate in food webs, and achieve highest concentrations in marine mammals and birds. PFCs are of particular ecological and toxicological concern due to their tendency to biomagnify in food webs and cause adverse health effects, including reproductive damage, immunotoxicity, and hepatotoxicity [18]. Of further interest is the unique physicochemical nature of PFCs. Whereas many legacy POPs are lipophilic and therefore accumulate in fatty tissues, PFCs tend to accumulate primarily in protein-rich tissues, such as the liver. Two PFAAs, perfluorooctane sulfonate (PFOS) and perfluorooctanoate (PFOA), represent the most commonly investigated PFCs of significant risk to wildlife and humans due to their ubiquitous nature, global fate and transport, high biomagnification potential, and toxicity risks, especially in aquatic and marine food webs of the northern hemisphere and Arctic [18–21]. Phase out

programs designed to eliminate the production of PFOS were established for some regions in the early 2000s, followed by the addition of PFOS to the list of restricted POPs under the Stockholm Convention on Persistent Organic Pollutants in 2009 [8]. Despite these initiatives, production of PFOS, PFOA, and several other PFAAs still take place around the world, including several developing countries [22–25]. One of the priority actions under the Antarctic Treaty is the assessment and monitoring of POPs, including PFCs, in Antarctica. Considering that assessments of PFCs in the Antarctic are limited, questions linger about the long-range environmental transport of these substances to the Southern Hemisphere, and the capacity of these substances to bioaccumulate in Antarctic food webs.

PFCs have been detected in various Antarctic environmental media and biota, typically in the pg/g to ng/g range, though many samples return nondetectable levels or levels below the minimum level of quantification [9]. Recent studies show that levels of many PFCs in Arctic environments have been increasing, with concentrations of several PFCs equivalent to or surpassing that of DDT, PCBs, PBDEs, and other organochlorine pesticides [19]. Similar patterns are anticipated for PFCs in Antarctic environments as they are continuously delivered from other geographic locations via long-range transport.

Although some PFCs are already categorized as POPs, the majority of these substances are not subject to global or local controls. To ensure that potential impacts of pollutants on Antarctic wildlife are considered in the global environmental agenda and throughout negotiations on commercial chemical production and use, it is important that a high-quality research program is developed on the fate and effects of contaminants in Antarctic ecosystems and wildlife. As part of an ongoing scientific initiative and collaboration between the Ecuadorian Antarctic Institute (INAE), Simon Fraser University (Canada), the Institute of Ocean Sciences (IOS, Fisheries and Oceans Canada, DFO), and the Escuela Superior Politecnica del Litoral (Ecuador), a study to investigate and monitor PFCs was initiated in Peninsula Antarctica around the surrounding areas of the Ecuadorian Station “Pedro Vicente Maldonado” during the 2009 Ecuadorian–Antarctic expedition. In this chapter, we provide one of the primary findings on PFCs in sediments and biotic matrices, including lichens as well as feces and feathers from the southern giant petrel (*Macronectes giganteus*) and gentoo penguin (*Pygoscelis papua*), and evaluate the use of noninvasive techniques to monitor emerging organic contaminant of concern in the Antarctic environment.

2. Materials and methods

2.1. Study Area and sampling

The Ecuadorian Research Station “Pedro Vicente Maldonado” (Maldonado Station, hereafter) is located at Fort William Point, Greenwich Island (62°31'S; 59°46'W; Figure 1). The study area encompassed the Barrientos Island (62°24'01"S; 59°43' 52"W), Dee Island (62°25'48.5" S; 59°47'69.6" W), Punta Ambato (62°26'33" S; 59°47'28.8" W), and the surroundings of the Maldonado Station (62°27'59"S; 59°43'32.5"W), as illustrated in Figure 1. Sampling was conducted using three tracks established by the Maldonado Station to access the coastline of

Fort Williams, which enclose two sampling zones: Ensenada Guayaquil and Bahia Chile. These sectors are only used by technical and military personnel that work at the Station and visiting scientists that come to the island for research purposes. Barrientos Island is used principally as a tourist stopover for cruise ships where tourists land and walk around the island for bird-watching. In Dee Island and Punta Ambato, sampling was deployed around the coastline. All sampling was done during the Austral summer and seabird breeding seasons of 2009. The collection of abiotic and biotic samples is described as follows.

2.1.1. Sediments

Sediment samples were collected from three locations in the Antarctic Peninsula including Dee Island ($n = 1$ site), Maldonado Station ($n = 2$ sites), and Punta Ambato ($n = 2$ sites) (Figure 1). Sediment samples were directly collected using 100 mL centrifugation tubes, stored at $< 4^{\circ}\text{C}$ until transportation to the laboratory in Canada.

2.1.2. Seabirds

Gentoo penguins (*Pygoscelis papua*) and southern giant petrels (*Macronectes giganteus*), two species of seabirds that inhabit the Antarctic Peninsula, were identified as potential bio-indicators of PFCs contamination. The main reason for selecting seabirds is based on studies showing that bird populations are most affected by contaminants, specifically POPs, among wildlife species (see [26] for a review). Bird species have the greatest capacity to biomagnify chemicals because of their highly energy-efficient metabolic system and also because of their high trophic position within the food web. Bird populations are therefore often at high risk from bioaccumulative substances, and can act as the “canary in the coal mine” for the larger Antarctic ecosystem.

In this context, we conducted a noninvasive sampling technique to minimize or completely avoid the impacts of lethal or invasive sampling on the local bird populations. Sampling focuses on the collection of shed/molting feathers and excreted fecal matter from nesting sites. Because of the very high affinity of PFCs for protein, feathers are good noninvasive sampling media for PFCs, as they consist mainly of protein matter (i.e., keratin, a high molecular weight protein). Feathers have also been used to successfully monitor mercury in seabird populations such as brown skuas, *Catharacta lonnbergi*, chinstrap penguins, *Pygoscelis antarctica*, and gentoo penguins, *P. papua*, in our study area [27], as well as PFCs in the feathers of aquatic and marine birds, including grey heron (*Ardea cinerea*) and herring gull, (*Larus argentatus*) from the Northern Hemisphere [28]. Fecal matter is known to contain some of the highest concentration of contaminants due to the gastrointestinal magnification that occurs in the intestinal tract of consumer organisms. In addition, the contaminant concentrations in fecal matter are related to compounds absorbed by the organism, such that they can provide a measure of accumulated concentrations. The low capacity to migrate to the gaseous phase (i.e., air) and high octanol-air partition coefficient (K_{OA}) of the analytes (Table 1) cause minimal losses of the contaminants from feces or feathers to the air after feathers or fecal matter have been dropped. This means that the concentrations of the chosen analytes can remain a measure of bird exposure levels long after the feces have been excreted or feathers have been shed.

Molted feathers were collected randomly in and around nests and colonies of petrels surrounding the Maldonado's Station and stored in ziploc-type plastic bags ($n = 5$). Only one bag of feather samples for gentoo penguin was collected from the Maldonado's Station. Fecal matter samples from gentoo penguins were collected from nesting sites and colonies around the Maldonado's Station ($n = 9$), Barriento Island ($n = 7$), and Dee Island ($n = 3$). All feces samples were placed into 20 mL glass vials. Both feather and fecal samples were stored in coolers and transported by airplane with dry ice (-20°C) until transportation to the lab in Canada.

2.1.3. Lichens

Lichen (*Usnea aurantiaco-atra*) samples ($n = 5$) were collected from rocky areas around the surroundings of the Maldonado's Station and wrapped with clean, sterile aluminum foil and stored in ziploc plastic bags until further transportation to the lab. The rationale to select lichens is based on the premise that this biological matrix can be used as a potential monitor and indicator of global atmospheric transport of some PFCs to the Antarctic Peninsula.

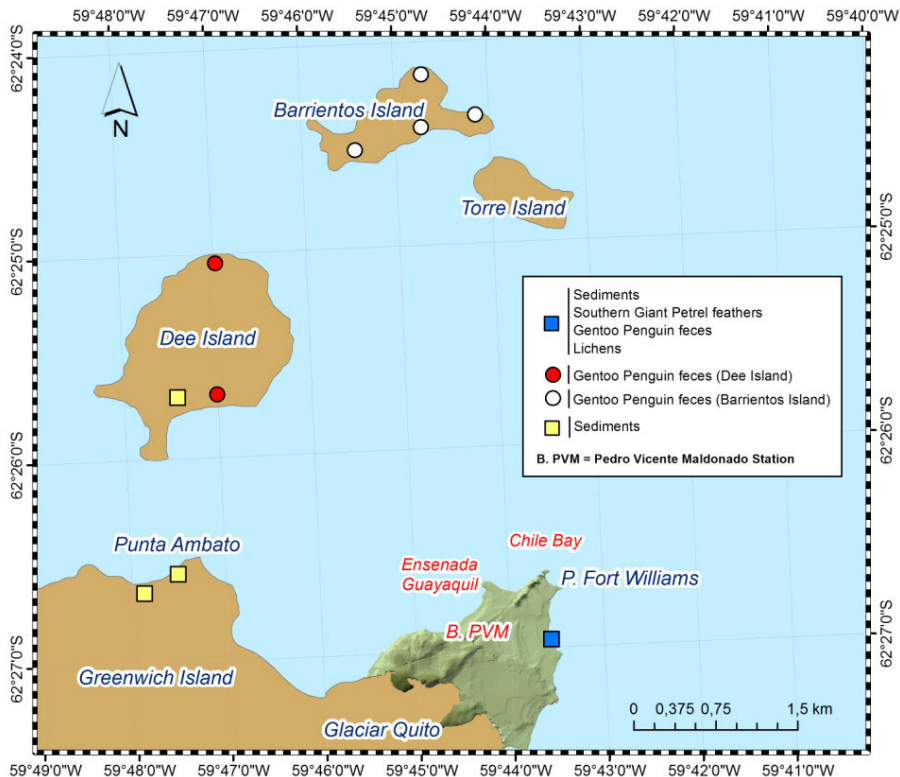


Figure 1. Geographical location of the study area and sampling sites in the Antarctic Peninsula

2.2. PFC physical – Chemical properties

Table 1 summarizes the compiled physical–chemical properties for the various PFCs studied, including molecular weights (MW), log octanol–water partition coefficients (log K_{OW}), log octanol–air partition coefficients (log K_{OA}), and log D values. Because the physicochemical properties of PFCs are considerably different from that of many other legacy POPs (i.e., they can be ionized at environmentally relevant pH), it is important to recognize that relationships applicable to other POPs may be less relevant when applied to PFCs and other ionizable compounds. For instance, many organic compounds of concern, including numerous agricultural and pharmaceutical compounds, are lipophilic in nature, and will tend to accumulate in fatty tissues [3]. The octanol–water partition coefficient (K_{OW}) has become a common property used to describe the tendency of a substance to partition into lipid, as the behavior of octanol and lipid are quite similar. Octanol thus serves as a suitable surrogate for lipid, particularly within predictive bioaccumulation models [29]. However, K_{OW} describes the lipophilicity of neutral compounds, and is not necessarily applicable to ionizable organic compounds (IOCs) such as PFCs, where the measure of lipophilicity is pH-dependent [30]. Many PFCs are almost completely ionized at environmentally relevant pH [31]. A more applicable indicator for predicting the lipophilicity of ionizable substances is log D , where both the neutral and the ionic species of the compound are accounted for [30].

2.3. PFC analysis: Extraction and quantification

Sediment and biological samples were extracted and analyzed at the Institute of Ocean Sciences (IOS), Fisheries and Oceans Canada (DFO), Sidney, British Columbia, Canada. PFC concentrations were analyzed by liquid chromatography tandem mass spectrometry with double mass detectors (LC/MS/MS), as described elsewhere [19]. Analyte concentrations were determined with respect to the mass labeled quantification and internal standards using isotope dilution method. Fifteen PFCs were examined in this study (Table 1). High purity (>95%) analytical standards, including perfluorobutane sulfonic acid (PFBS), perfluorohexanesulfonic acid (PFHxS), PFOS, perfluorobutanoic acid (PFBA), perfluoropentanoic acid (PFPeA), perfluorohexanoic acid (PFHxA), perfluoroheptanoic acid (PFHpA), PFOA, perfluorononanoic acid (PFNA), perfluorodecanoic acid (PFDA), perfluoroundecanoic acid (PFUnA), perfluorododecanoic acid (PFDoA), perfluorotetradecanoic acid (PFTA), and perfluorooctanesulfoamide (PFOSA), were used. Mass-labeled internal standards included six PFCs ($^{13}\text{C}_2$ PFOA, $^{13}\text{C}_2$ PFDA, $^{13}\text{C}_2$ PFDoA, and $^{13}\text{C}_4$ PFOS, $^{13}\text{C}_4$ -PFOA). Calibration curves were constructed from the analysis of calibration standard solutions (range 0.08–5.0 ng/mL).

Various calibration standards and standard additions were prepared and used as quality assurance/quality control (QA/QC). QA/QC measures included initial method validation work, consisting of analyte recovery experiments of native PFCs in clean sediments and biota. The method of detection limit (MDL) was set equal to the concentration of the method's level of quantification (MLOQ) for samples and subtracted from quantified concentrations of each analyte (Table 2). Only corrected data above the MLOQ are reported

in this work. Concentrations of PFCs were expressed on a wet weight basis (ng/g ww). Extraction methods are briefly described as follows.

2.3.1. Sediment

Sediment samples (≈ 10 g wet weight) were added to 50 mL polypropylene centrifuge tubes and spiked with internal surrogate spiking solution (360 ng of $^{13}\text{C}_2$ PFOA, 120 ng of $^{13}\text{C}_2$ PFDA, 120 ng of $^{13}\text{C}_2$ PFDoA, and 120 ng of $^{13}\text{C}_4$ PFOS; Table 1). After 20 min, 10 mL of 0.1% acetic acid in MeOH was added, and samples were extracted on a shaker table for 16 h. After extraction and centrifugation, 1 mL was pipetted into 1.5 mL Ependorf vial containing 25 mg of activated carbon. Then the vial was subject to centrifugation for 30 min at 14,000 rpm; 300 μL of supernatant was taken and combined with 300 μL of water and 50 μL of 20 ppb of recovery standard and centrifuged again for 15 min at 14,000 rpm. Then, 300 μL of supernatant was used for LC/MS/MS analysis (i.e., injection volume=100 μL for LC/MS/MS).

2.3.2. Feathers

Approximately 0.74 g of feather was weighed, and then homogenized by adding first HNO_3 (e.g., 4×0.9 mL, 2×0.9 mL) with a series of vortexing steps until the whole particulates completely disappeared within 3 h. Samples were set up for digestion at room temperature (RT) for 12 h. Afterwards, 15 mL of 5 M NaOH prepared in water was added to samples and shaken on a shaker table for 5 min. The pH was measured to ensure the sample was acidic enough (i.e., pH~3–4) prior to direct injection in LC/MS/MS (large volume injection). After neutralization and extraction with 2.5 mL MeOH for a total volume 27.5 mL, ion suppression was found from recoveries; therefore, additional dilution (10 \times) was done until ion suppression was reduced (i.e., injection volume = 200 μL).

2.3.3. Feces

Penguin fecal matter (~ 0.65 g of feces) was weighed and homogenized with HNO_3 (e.g., 4×0.9 mL, 2×0.9 mL and vortexing). Samples were set at RT for digestion during 12 h. After digestion, samples were neutralized to pH equal to 3.2–4.2, brought up to 50 mL and centrifuged at 6000 rpm for 25 min; 100 μL of recovery standard ($^{13}\text{C}_4$ -PFOA) was added to an aliquot of 400 μL and injected into LC/MS/MS (i.e., injection volume= 200 μL).

2.3.4. Lichen

Lichen (2 g) was extracted based on the methodology described in reference [32]. After extraction, 4 mL of solution was blown down to 2 mL, followed by collecting 1 mL aliquot and added into 1.5 mL Ependorf vial containing 25 mg of activated carbon. The vial was subject to centrifugation for 30 min at 14,000 rpm and 300 μL of supernatant was obtained and combined with 300 μL water and 50 μL of 20 ppb recovery standard and centrifuged for 15 min at 14,000 rpm. Then, 300 μL of supernatant was used for LC/MS/MS analysis (i.e., injection volume=100 μL).

Chemical Name	Abbreviation	Formula	MW (g/mol)	Log* K_{OW}	Log* K_{OA}	Log** D
Target Analytes						
Perfluorobutanoic acid	PFBA	C ₃ F ₇ COO-	214.0	1.3	5.0	0.060
Perfluoropentanoic acid	PFPeA	C ₄ F ₉ COO-	264.0	2.1	5.3	0.54
Perfluorohexanoic acid	PFHxA	C ₅ F ₁₁ COO-	314.1	3.1	5.6	1.1
Perfluoroheptanoic acid	PFHpA	C ₆ F ₁₃ COO-	364.1	2.8	5.9	1.6
Perfluorooctanoic acid	PFOA	C ₇ F ₁₅ COO-	414.1	3.6	6.3	2.3
Perfluorononanoic acid	PFNA	C ₈ F ₁₇ COO-	464.1	4.5	6.6	2.9
Perfluorodecanoic acid	PFDA	C ₉ F ₁₉ COO-	514.1	5.4	6.8	3.5
Perfluoroundecanoic acid	PFUnA	C ₁₀ F ₂₁ COO-	564.1	6.4	7.1	4.2
Perfluorododecanoic acid	PFDoA	C ₁₁ F ₂₃ COO-	614.1	7.1	7.4	5.0
Perfluorotetradecanoic acid	PFTA	C ₁₃ F ₂₇ COO-	714.1	8.8	8	6.1
Perfluorobutane sulfonic acid	PFBS	C ₄ F ₉ SO ₃ -	300.1	N/A	N/A	-0.53
Perfluorohexane sulfonic acid	PFHxS	C ₆ F ₁₃ SO ₃ -	400.1	N/A	N/A	0.54
Perfluorooctane sulfonic acid	PFOS	C ₈ F ₁₇ SO ₃ -	500.1	4.3	7.8	1.7
Perfluorodecane sulfonic acid	PFDS	C ₁₀ F ₂₁ SO ₃ -	600.1	N/A	N/A	3.1
Perfluorooctane sulfonamide	PFOSA	C ₈ F ₁₇ SO ₂ NH ₂	499.1	6.3	8.4	-
Mass Labeled Standards						
Perfluorooctanoic acid ^a	¹³ C ₂ -PFOA	-	-	-	-	-
Perfluorodecanoic acid ^b	¹³ C ₂ -PFDA	-	-	-	-	-
Perfluorododecanoic acid ^c	¹³ C ₂ -PFDoA	-	-	-	-	-
Perfluorooctane sulfonic acid ^d	¹³ C ₄ -PFOS	-	-	-	-	-
Perfluorooctanoic acid ^e	¹³ C ₄ -PFOA	-	-	-	-	-

MW: molecular weight

*Log K_{OW} and K_{OA} values of individual PFCs were compiled from published values calculated using SPARC general partitioning model [33].

**Log D values were calculated at pH = 7.5 and T = 21°C using SPARC.

^a used to quantify PFBA, PFPeA, PFHxA, PFHpA, PFOA, PFNA

^b used to quantify PFDA and PFUnA

^c used to quantify PFDoA and PFTA

^d used to quantify PFBS, PFHxS, PFOS, PFDS, PFOSA

^e used to quantify recovery of mass labeled surrogates

Table 1. List of target perfluoroalkyl chemicals (PFCs) and radiolabeled surrogates monitored using LC/MS/MS.

Analyte	Sample Type			
	Feather	Feces	Lichen	Sediment
	MLOQ (ng/g) for 0.74 g sample	MLOQ (ng/g) for 2 g sample	MLOQ (ng/g) for 2 g sample	MLOQ (ng/g) for 10 g sample
PFBA	0.81	2.16	0.73	2.04
PFPeA	1.22	5.86	4.11	4.34
PFHxA	1.10	2.51	3.33	1.75
PFHpA	0.57	3.25	0.77	0.55
PFOA	1.14	3.08	3.34	2.34
PFNA	0.80	4.04	1.04	1.23
PFDA	0.82	2.80	0.51	0.40
PFUnA	2.78	4.01	0.78	0.17
PFDoA	1.79	7.10	0.22	0.26
PFTA	1.30	10.86	0.57	0.25
PFBS	0.03	0.92	0.09	0.12
PFHxS	0.10	0.45	0.19	0.25
PFOS	0.30	0.88	0.75	0.33
PFDS	2.44	1.42	0.12	0.13
PFOSA	1.43	N/A	0.76	0.32
FHUEA	0.64	2.96	0.19	0.25
FOUEA	1.06	1.55	0.08	0.27
FDUEA	4.02	4.73	0.31	0.25

Table 2. Method's limit of quantification (MLOQ) for PFC analytes measured by LC/MS/MS.

3. Result and discussion

3.1. PFC concentrations

Several PFC compounds showed concentrations above the MLOQ, as shown in Table 3. Perfluorotetradecanoic acid (PFTA), a chemical with a high K_{OW} and high K_{OA} that will persist for decades in humans, was measured in 60% of sediment samples, but undetected or below the MLOQ in lichens, feces, and feathers. Perfluoroheptanoic acid (PFHpA) was detected in all seabird feather samples (range = 1.60–2.85 ww ng/g; Table 3), and in 47% of penguin feces, ranging 0.37–22 ng/g ww. All lichen samples exhibited concentrations of perfluorohexanesulfonate (PFHxS), ranging 0.20–1.20 ng/g ww, while perfluorobutyric acid (PFBA), perfluoro-n-pentanoic acid (PFPeA), and PFHpA were measured in 80%, 60%, and 60% of lichen samples, respectively. PFOA and PFOS were not quantified in most samples (i.e., < MLOQ or ND; Table

3), except for the detection of PFOS in two penguin feces samples (2.8 and 3.14 ng/g ww), and PFOA in a single fecal sample (2.0 ng/g ww) and lichen (4.7 ng/g ww). The lack of PFOS and PFHxS detection in Antarctic seabird feathers contrasts with the levels of PFOS and PFHxS found in feathers of grey herons (PFOS: 247 ng/g dw; PFHxS: \approx 20 ng/g dw) and herring gulls (PFOS: 79 ng/g dw; PFHxS: > 30 ng/g dw) from the Northern Hemisphere (Flanders, Belgium) [28]. However, the absence of PFOA in our feather samples is consistent with the lack of detection of this compound in bird feathers from the same region [28]. For comparison purposes, the PFOA concentration detected in a sample of gentoo penguin feces was 14 times lower than the PFOA concentration (28.2 ng/g ww) detected in a single herring gull liver sample from Belgium [28]. Despite samples from other parts of the world that indicate a continued increase or no change in PFOS levels following the 2002 phase-out [34–37], a fast decline in PFOS concentrations has been observed in wildlife over the past decade [38,39]. PFDA, PFUnA, PFDoA, and PFOSA were not detected (ND) or < MDL. Except for the compound PFHpA, lack of detection of most analytes in samples and small sample sizes preclude undertaking robust statistical analyses for multisite or/and inter-species comparisons.

3.2. PFC patterns

Figure 2 shows the composition of PFCs observed in biotic and abiotic samples. PFHpA was the only compound detected in feathers of both petrels and penguins, accounting for 100% of total PFCs, while PFTA was equal to 100% of PFCs in sediment samples (Figure 2). PFHpA was also found in feces and lichen samples making up 24.5% and 23% of total PFCs, respectively. PFDS contributed to 54% and 17% of PFCs in feces and lichens, contrasting with PFPeA and PFHxS, which accounted for 3.4% and 50%, and for 4% and 21% of PFCs in feces and lichen samples, respectively. PFNA accounted for 38% of the PFCs in feces. These patterns clearly show that both perfluorinated sulfonates (PFSAs) and carboxylates (PFCAs) exhibit different fractions in seabirds, reflecting the potential role of biotransformation in shaping the accumulation of these compounds.

3.3. Bioaccumulation of PFCs

The biomagnification factor (BMF) [40] for PFHpA (i.e., $BMF = C_B/C_D$, where C_B is the PFHpA concentration detected in the predator, the giant petrel, and C_D is the PFHpA concentration observed in the diet/prey, gentoo penguin) was calculated using feather concentrations, as this was the only PFC compound readily detected in 100% of feathers samples. Hence, the concentrations of PFHpA in the petrel feathers (i.e., mean \pm SD = 2.6 ± 0.60 ng/g ww; $n = 5$) and that of the penguin (1.60 ng/g ww; $n = 1$; Table 3) were used as surrogates for concentrations in the tissues of the whole organism, assuming that the birds had been exposed to the compound for a sufficiently long time to allow the concentrations to reach steady state [40]. The criterion applied to indicate that PFHpA was biomagnified in petrels was a $BMF > 1$, such that a BMF greater than 1 indicates that the chemical is a bioaccumulative substance [41]. Here, we found that the BMF was close to 2 (i.e., 1.6), indicating that PFHpA biomagnifies in petrels. Although the concentrations of PFHpA in feces appear to be relatively higher than the concentrations found in lichen and feathers, comparisons of the PFHpA concentrations among

Sample Type	Sample code	PBFA	PFBA	PFHA	PFHpA	PFNA	PFDA	PFUnA	PFDoA	PTA	PBFS	PFHS	PFOS	PFDS	PFOSA	PFHUA	FOUA	FUUA
Feather	PETFEA1202MS1	ND	ND	ND	2.85	ND	ND	ND	ND	ND	ND	ND	<2.44	ND	ND	ND	ND	ND
	PETFEA1202MS2	ND	ND	ND	2.32	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
	PETFEA1202MS3	ND	ND	ND	3.54	ND	ND	ND	ND	ND	<0.03	ND	<2.44	ND	ND	ND	ND	ND
	PETFEA1202MS4	ND	ND	ND	2.24	ND	ND	ND	ND	ND	ND	ND	<0.30	ND	ND	ND	ND	ND
	PETFEA1202MS5	ND	ND	ND	1.93	ND	ND	ND	ND	ND	ND	ND	<0.30	<2.44	ND	ND	ND	ND
	GENFEA1002	ND	ND	ND	1.63	ND	ND	ND	ND	ND	ND	ND	<2.44	ND	ND	ND	ND	ND
	GF1280/D12	N/A	1.95	<2.51	<3.25	ND	<4.0	ND	ND	ND	1.42	4.90	<0.88	ND	ND	11.7	34.4	ND
	GF280/D11	N/A	ND	<2.51	<3.25	<3.1	9.50	ND	ND	ND	ND	<0.45	<0.88	<1.42	ND	ND	ND	ND
	GF280/D13	N/A	ND	<2.51	10.2	<3.1	9.50	ND	ND	ND	ND	0.33	<0.45	3.33	ND	ND	ND	ND
	GFNFEC020MS1	N/A	2.63	0.94	ND	<3.1	3.90	ND	ND	ND	ND	ND	<0.88	108	ND	ND	32.7	<4.7
Feces	GFNFEC020MS2	N/A	ND	3.37	1.15	<3.1	ND	ND	ND	ND	0.28	<0.45	<0.88	<1.42	ND	ND	ND	ND
	GFNFEC050MS1	N/A	ND	2.30	4.38	<3.1	ND	ND	ND	ND	2.39	2.67	2.80	117	ND	ND	37.8	0.56
	GFNFEC050MS2	N/A	ND	3.27	0.37	<3.1	ND	ND	ND	ND	ND	ND	ND	10.5	ND	ND	ND	ND
	GFNFEC050MS3	N/A	ND	<2.51	0.16	<3.1	N/A	ND	ND	ND	0.32	<0.45	ND	ND	ND	ND	ND	ND
	GFNFEC050MS4	N/A	ND	ND	ND	<3.1	N/A	ND	ND	ND	ND	ND	ND	<1.42	ND	ND	ND	ND
	GFNFEC050MS5	N/A	9.65	1.33	11.09	2.03	7.23	ND	ND	ND	1.52	1.77	3.14	130	ND	ND	37.9	0.07
	GFNFEC310B1	N/A	ND	<2.51	<3.25	<3.1	1.23	ND	ND	ND	ND	ND	ND	<1.42	ND	ND	ND	ND
	GFNFEC310B2	N/A	ND	ND	<3.25	<3.1	5.50	ND	ND	ND	ND	<0.45	<0.88	<1.42	ND	ND	ND	ND
	GFNFEC310B3	N/A	ND	ND	21.9	<3.1	1.23	ND	ND	ND	ND	ND	<0.88	<1.42	ND	ND	ND	ND
	GFNFEC310B4	N/A	ND	ND	ND	<3.1	5.01	ND	ND	ND	0.38	ND	ND	<1.42	ND	ND	ND	ND
Lichen	GFNFEC310B5	N/A	ND	ND	<3.25	<3.1	5.81	ND	ND	ND	ND	ND	23.1	ND	ND	ND	ND	ND
	GFNFEC310B6	N/A	ND	ND	<3.25	<3.1	4.0	ND	ND	ND	ND	<0.45	<0.88	<1.42	ND	ND	ND	ND
	GFNFEC310B7	N/A	9.65	<2.51	2.41	<3.1	N/A	ND	ND	ND	ND	ND	117	ND	ND	32.7	<4.7	ND
	PENFEC260MS1	N/A	3.80	<2.51	17.61	<3.1	<4.0	ND	ND	ND	ND	ND	<0.88	75.5	ND	ND	32.4	<4.7
	PENFEC290MS2	N/A	ND	ND	ND	<3.08	ND	ND	ND	ND	0.54	ND	ND	ND	ND	ND	ND	ND
	L1002MS1	2.91	4.06	ND	5.28	ND	ND	ND	<0.22	<0.57	<0.09	1.05	ND	<0.12	<0.76	<0.19	ND	ND
	L1002MS2	<0.73	<4.1	ND	ND	ND	<1.04	ND	<0.22	<0.57	0.80	0.80	ND	0.32	<0.76	<0.19	ND	ND
	L1002MS3	0.55	6.05	ND	1.77	<3.3	<1.04	ND	<0.22	<0.57	<0.09	0.11	<0.75	ND	<0.76	<0.19	ND	ND
	L1002MS4	1.30	5.29	<3.3	0.87	4.69	<1.04	ND	<0.22	<0.57	ND	1.16	<0.75	ND	<0.76	<0.19	ND	<0.31
	LB2601MS1	1.22	<4.1	<3.3	ND	<3.3	<1.04	ND	<0.22	<0.57	ND	0.19	<0.75	0.07	<0.76	<0.19	ND	<0.31
Sediment	Control	ND	ND	ND	ND	<1.2	<0.40	<0.17	<0.26	0.12	ND	ND	<0.33	ND	ND	ND	ND	<0.25
	S280/D11	ND	ND	ND	ND	<2.3	<0.40	<0.17	<0.26	ND	ND	ND	<0.33	<0.13	<0.32	<0.25	<0.27	<0.25
	S1902MS1	ND	ND	ND	ND	<2.3	<0.40	<0.17	ND	ND	ND	ND	<0.33	<0.13	<0.32	<0.25	<0.27	<0.25
	S1602MS2	ND	ND	ND	ND	<2.3	<0.40	<0.17	0.14	ND	ND	<0.25	<0.13	<0.32	<0.32	<0.25	<0.27	<0.25
	S2701PA1	<2.04	ND	ND	ND	<2.3	<0.40	<0.17	<0.26	0.11	ND	ND	<0.13	<0.32	<0.32	<0.25	<0.27	<0.25
S2701PA2	ND	<4.3	ND	ND	<2.3	<0.40	<0.17	<0.26	0.11	ND	ND	<0.13	<0.32	<0.32	<0.25	<0.27	<0.25	

ND means that the compound of interest is not detectable with the current method; no visible peak in the ion chromatogram.
 Values showing the less than sign (<) are equal to <MLOQ (see Table 2). <MLOQ means that the compound of interest is detectable with the current method, but unable to quantify due to the low concentration in the sample; there is a visible peak in the ion chromatogram, but not quantifiable.
 N/A means not applicable, identification uncertainly due to chromatographic interference.

Table 3. Quantification data of PFCs (ng/g ww) in feather, feces, lichen, and sediment samples collected in the Antarctic Peninsula. Data taken and modified from reference [63].

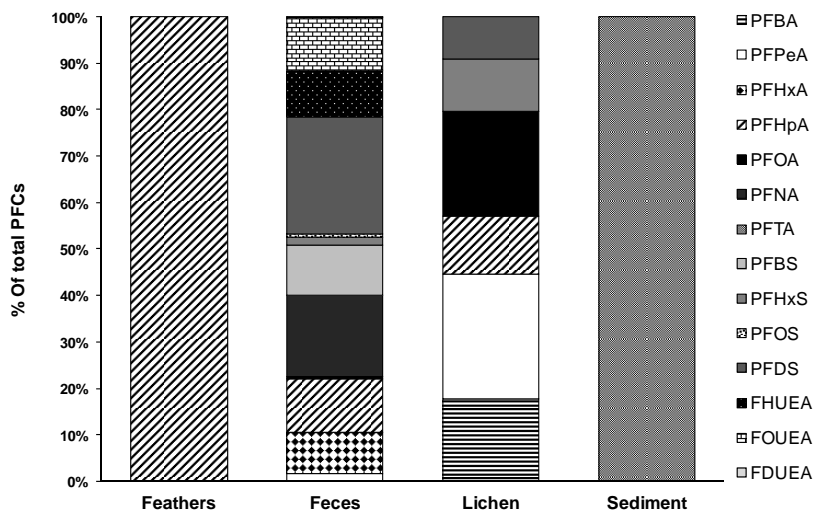


Figure 2. Composition pattern of PFC compounds detected in biotic (feathers and feces of seabirds, and lichen) and abiotic (sediment) samples from the Antarctic Peninsula. Abbreviations for PFC chemical names are defined in Table 1.

biota samples show lack of significant differences (Welch's ANOVA, $p > 0.05$; Tukey–Kramer HSD (honest significant difference) test, $p > 0.05$), as shown in Figure 3.

To further illustrate the behavior of PFC concentrations in these samples, detected PFC compounds were plotted as a function of $\log D$ and $\log K_{OA}$, as shown in Figure 4. The majority of PFCs concentrations observed in biotic samples (i.e., feces and lichens) fall within $\log D$ values between 0 and 3, as seen in Figure 4A. While concentrations of PFCs tend to increase with increasing $\log D$ values from $\log D$ of 0 to $\log D$ of 3 in feces, PFC concentrations appear to decrease as the $\log D$ increases within the same range of $\log D$ values in lichens (Figure 4A). This observation may be an indication that both ionized and unionized forms of PFC compounds with low $\log D$ values (i.e., PFBA, PFPeA, PFHxA, PFHxS, PFHpA, PFOS, PFOA, PFNA, PFDS) are present in some organisms residing in this region and prone to potential transportation by oceanic currents (e.g., Antarctic Circulation Current) from either continental/regional or local sources (i.e., international military bases and research stations) to the Antarctic Peninsula. Similarly, most PFCs concentrations observed in these samples, especially in lichens, fall within $\log K_{OA}$ values of 5.0 and 6.5 (Figure 4B). Although concentrations for some PFC compounds show a tendency to decrease with increasing $\log K_{OA}$ in feces (i.e., PFBS, PFHxS, PFPeA, PFHxA, PFHpA, PFOA, PFNA, PFOS), concentrations for similar PFCs seem to increase as $\log K_{OA}$ increases in lichens (i.e., PFDS, PFHxS, PFBA, PFPeA, PFHpA, PFOA), as seen in Figure 4B. These trends may support the notion that low molecular weight compounds (e.g., 214–414 g/mol) with low $\log K_{OA}$ are likely to be subject to long-range atmospheric transport and potentially reaching the region, where these compounds accumulate in biotic compartments, mainly in natural air samplers such as lichens and secondary in air-breathing organisms such as seabirds (petrels, penguins).

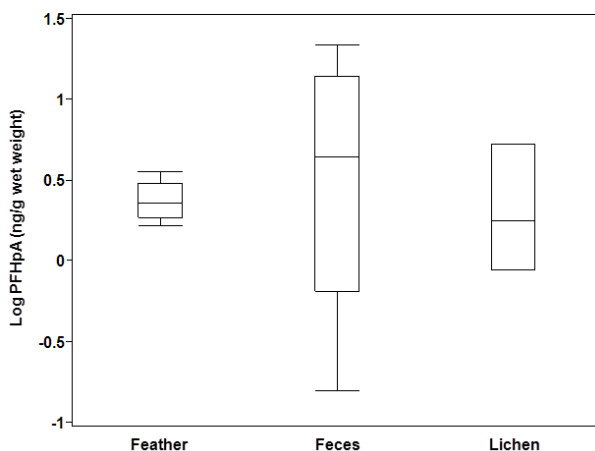


Figure 3. Box plots showing log transformed concentrations of PFHpA (ng/g ww) detected in lichen ($n=3$), and feather ($n=6$) and feces ($n=9$) of seabirds from the Antarctic Peninsula. The internal line across the box is the median; the ends of the box are the 25% and 75% quartiles; and the whisker bars are the minimum and maximum values. Because of unequal variances (i.e., heteroscedasticity; Bartlett test, $p < 0.005$), a Welch's ANOVA, followed by a Tukey–Kramer HSD test, was used for the multicomparison, showing no significant differences in PFHpA concentrations among the biotic samples ($p > 0.05$).

3.4. PFC health risks

Concentrations of PFCs detected in feathers and feces of the two seabird species studied here are well below the toxicological reference value (TRV) of PFOS (600 ng/g ww), calculated as an exposure threshold value for birds in nature, especially for apex avian predators [42]. This comparison indicates that gentoo penguins and petrels are not at risk by PFOS toxic effects.

3.5. Transport mechanisms, global and local sources of PFCs to the Antarctic

There is still a degree of uncertainty surrounding the dominant pathway of PFC movement to the Antarctic, though researchers have highlighted two primary mechanisms generally accepted as the major modes of PFC transportation to the Antarctic: atmospheric and oceanic. Neutral, volatile precursor compounds, such as perfluorinated sulfonamide alcohols (FOSEs), perfluorooctane sulfonamides (FOSAs), and fluorotelomer alcohols (FTOHs), referred to as “flyers”, are capable of being delivered to the Antarctic via fast, direct transport of contaminated wind, as opposed to cold trapping, as is common for many legacy POPs [12, 43–45]. Following deposition, these compounds are degraded via oxidation to form ionic PFCs, including PFSA and PFCA [12, 44–48]. Evidence supporting this mechanism of travel includes measurements of FTOHs from Europe to the Antarctica showing declining concentrations in the atmosphere with increased distance from sources in the Northern Hemisphere [12]. Given the far distances PFCs must travel to reach the Antarctic, in combination with short atmospheric residence times (ranging on average from 10 to 50 days), the level of effectiveness associated with atmospheric delivery of PFCs is relatively low. Additionally, the yield of ionic

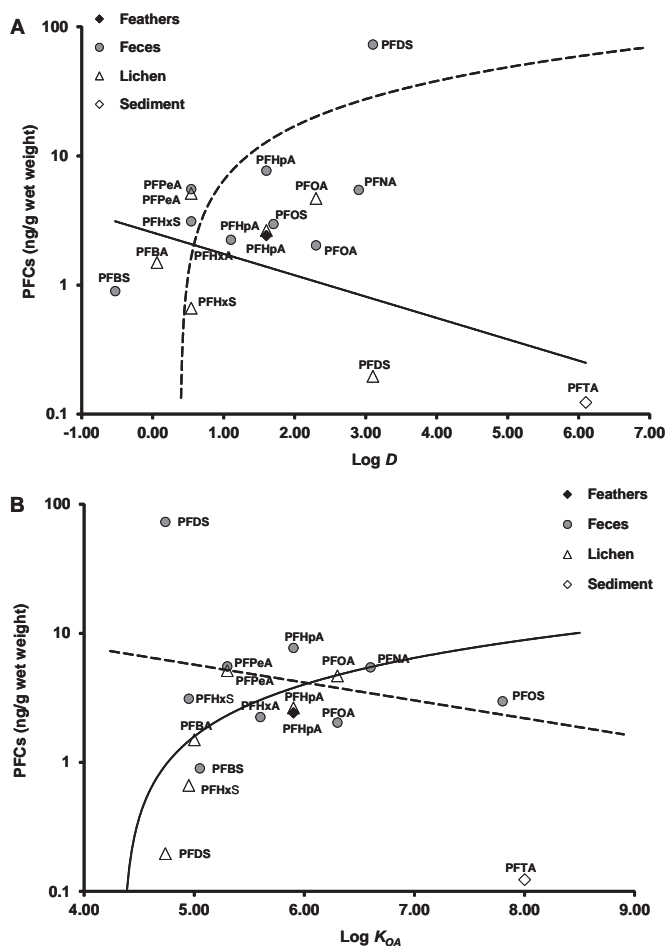


Figure 4. Concentrations of PFC compounds (logarithmic scale in ng/g wet weight) measured in sediments, lichens, and seabird feces and feathers from the Antarctic Peninsula as a function of $\log D$ (A) and $\log K_{OA}$ (B). In Figure 4A, the solid line shows the behavior of detected PFC concentrations versus $\log D$ in lichen; the dashed line indicates the trend of detected PFC concentrations versus $\log D$ values in feces. In Figure 4B, the solid line indicates the behavior of detected PFC concentrations versus $\log K_{OA}$ in lichen, while the dashed line shows detected PFC concentrations versus $\log K_{OA}$ in feces. Abbreviations for PFC chemical names are defined in Table 1.

PFCs produced via oxidation of precursor compounds once transported to the Antarctic is often low [10, 11, 43–45, 47].

It is therefore expected that most PFCs are delivered to the Antarctic in their ionic, water-soluble state via the oceans [12, 14, 49]. Oceanic transport functions on a slower time scale for the Antarctic (in the order of decades, compared to days or weeks for atmospheric transport) because of the circulation patterns of the Southern Ocean, protecting the Antarctic from immediate fluxes in PFC concentrations as they are released elsewhere in the world. As time

progresses, however, contamination from oceanic sources is anticipated to increase [10, 11, 45]. Slow oceanic transport is cited as the reason for increasing PFC concentrations in the Arctic since the 1950s [14]. Models designed for Arctic research show that if oceanic transport to the Arctic ceased, the quantity of PFCs and their precursors delivered to the Arctic via atmospheric transport could not account for the concentrations measured in water, and thus marine transport is considered to be more important than atmospheric transport [15, 43]. It is also important to note that atmospheric and oceanic transport may be difficult or impossible to discern. For instance, PFCs found in the ocean are made up of three inputs: direct emissions to water, atmospheric deposition into water, and precursor compounds into water followed by degradation to ionic PFCs [14].

Among the compounds found in this study, PFHpA, PFBA, and PFPeA are byproducts of stain/grease-proof coatings on food packaging, couches, and carpets, while PFHxS was used in fire-fighting foams and carpet treatments and phased out of consumer products along with PFOS and PFOA by the major manufacturer (3M Co.) in the early 2000s due to health risks. While long-range atmospheric and oceanic transport of PFCs may partially explain the ubiquitous nature of these contaminants in the Antarctic Peninsula, military bases and infrastructure of nations established there may also be contributing sources of PFCs in Antarctic ecosystems. Atmospheric long-range transport of PFAAs as marine aerosols and degradation of PFCA and PFSA precursors such as low molecular weight FTOHs and acrylates/acids (FTAs) or perfluoroalkyl sulfonamids (FASA) and sulfonamido ethanols (FASE), which are more volatile and released to the atmosphere during fluoropolymer production processes, can be considered as other major pathways [11, 14, 44, 50, 51] to reach and deposit on the Antarctic Peninsula.

Additional and potential sources of PFAAs in the Antarctic Peninsula include aqueous film-forming foams (AFFF) and emissions of a current use insecticide, sulfluramid (N-ethyl perfluorooctane sulfonamide), to control leaf-cutting and fire ants in South America [20, 52; J. Benskin, pers. comm., June 2012). AFFF formulations have consisted of perfluoroalkyl sulfonates (PFHxS, PFOS, PFDS) and more recently, fluorotelomer sulfonamide-based surfactants. While these latter materials can degrade down to short-chain perfluoroalkylcarboxylates (typically C4, C5, C6 PFCAs), sulfluramid can degrade to PFOS, FOSA, and PFCAs by abiotic and/or biological processes [53, 54]. Sulfluramid is manufactured in Brazil (≈ 30 tons/year in 2007), and, in 2006, about 12 tons was exported to 13 other Central and South American countries [23, 55]. Because this insecticide is a semivolatile substance, it could be transported atmospherically to the Antarctic. Sulfluramid degradation products include PFOSA, PFOS, and potentially PFOA [52]. Despite high concentrations of PFOS, PFOA, and PFOSA measured off the Atlantic coast of South America (South Atlantic), increasing from Brazil to near Rio de la Plata (Argentina–Uruguay), attributed to the use of this substance [52], PFOS and PFOA were not detected in these Antarctic samples, with the exception of two feces samples and a lichen sample. This indicates that these two compounds have not yet fully reached the Antarctic Peninsula region, or local sources are not significant. The detection of several PFCA compounds in the present study is of particular importance as increasing trends of PFCA precursors (i.e., FTOHs) was observed in the Arctic with doubling times of 2.3–3.3 years from 2006 to 2012 [6].

4. Impact assessment, environmental management, and monitoring implications

The Ecuadorian Pedro Vicente Maldonado Scientific Station has been operated since 1988 shortly after Ecuador signed the Antarctic Treaty System (ATS) in 1987. In 1988, Ecuador became an associated member of the Scientific Committee for Antarctic Research (SCAR), and in November 1990 became a consultative member of the ATS [56, 57]. To accomplish this task, Ecuador fulfilled the Antarctic Treaty of “peaceful purposes” and “freedom of scientific investigation” [58]. The commitment to the protection of the Antarctic environment requires being in compliance with the Madrid Protocol, which since 1991 is the prime basis for environmental management of the Antarctic terrestrial and near-shore environments. At the Maldonado Station, the Antarctic environmental management program deploys and integrates a range of generic and international tools, including environmental impact assessments (EIAs), monitoring of pollutants in the marine environment, species and habitat protection, following the environmental principles of the Madrid Protocol, and the administrative and procedural mechanisms of the Committee for Environmental Protection (CEP) [58]. The Ecuadorian Antarctic Institute (INAE) has established good environmental practices, and trained their staff and visitors with a conduct code according to the Madrid Protocol. During the 2010–2011 period, an EIA was performed by the INAE [59] to establish the baseline conditions of the military base and research station, including its area of influence, developing the Environmental Management Plan for the activities taken place at the Maldonado Station.

Considering that local activities and maritime traffic can pollute the surroundings of the Maldonado Station and Antarctic Peninsula, results from impact assessments and monitoring of water quality and potential contaminants have revealed the presence of other anthropogenic pollutants such as hydrocarbons and pesticides in the marine environment [59]. For instance, analyses of total hydrocarbons (THCs) were performed in water samples at the Guayaquil Bay, where concentrations ranged from 0.3 mg/L in sites near the Maldonado Station to 0.85 mg/L at Chile Bay [59]. Pesticide concentrations at several sites of Chile Bay revealed the presence of the organochlorine insecticides, including lindane or γ -hexachlorocyclohexane (γ -HCH) (i.e., 0.335 mg/L) and β -hexachlorocyclohexane (β -HCH or beta-BHC) (i.e., 0.00072 mg/L), as well as traces of the organophosphate malathion and the herbicide atrazine [59]. The long-range atmospheric transport associated with direct deposition or precipitation of volatile organic chemical is now recognized as a major pathway by which pesticides can be transported and deposited in surface waters and ice of Antarctica thousands of kilometers far from their sources [60–62]. Although relatively low concentrations of some PFCs were observed in biota and sediments samples of the remote western Antarctic Peninsula environment and local sources associated with scientific stations and military bases appear to not be significant sources of PFCs, this study gives further evidence of background concentrations around the Antarctic. Results from this study are consistent with research showing that volatile PFCs are subject to atmospheric long-range transport to remote regions, contributing to the contamination of persistent PFCA and PFSA compounds in the Antarctic [11, 44].

Despite the limited sampling and the need for replication to confirm the findings of this study, the biological (lichens, feathers, feces) and abiotic (sediments) samples assessed in this work can be used as environmental matrices to track the fate of PFCs at various temporal and spatial

scales in Antarctica. Of particular importance is the detection of several PFC compounds in seabird feces and lichens, which can be used as matrices for noninvasive sampling and long-term monitoring programs of PFCs in the Antarctic Peninsula. Long-term air monitoring and sampling of volatile PFCs in the Antarctica Peninsula is also recommended to elucidate atmospheric sources to the Maldonado's Station and surrounding environment.

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References

- [1] Vallack HW, Bakker DJ, Brandt I, Brostrom-Lunden E, Brouwer A, Bull KR, Gough C, Guardans R, Holoubek I, Jansson B, Koch R, Kuylenstierna J, Lecloux A, Mackay D, McCutcheon P, Mocarelli P, Taalman RDF. Controlling persistent organic pollutants – what's next? *Environ Toxicol Pharm* 1998;6:143–75.

- [2] Jones KC, de Voogt P. Persistent organic pollutants (POPs): state of the science. *Environ Pollut* 1999;100:209–21.
- [3] Mackay D, Fraser A. Bioaccumulation of persistent organic chemicals: mechanisms and models. *Environ Pollut* 2000;110:375–91.
- [4] Hansen KM, Christensen JH, Brandt J, Frohn LM, Geels C, Skjoth CA, Hedegaard GB. Overview of POPs in the environment: modeling their distribution and spread. *Simulation and Assessment of Chemical Processes in a Multiphase Environment*. NATO Science for Peace and Security Series C - Environmental Security. 2008; 255–72.
- [5] Tang H. Recent development in analysis of persistent organic pollutants under the Stockholm Convention. *Trend Anal Chem* 2013;45:48–66.
- [6] AMAP: Trends in Stockholm Convention persistent organic pollutants (POPs) in Arctic air, human media and biota. AMAP Technical Report to the Stockholm Convention. AMAP Technical Report No. 7 (2014); Oslo, Norway: Arctic Monitoring and Assessment Programme (AMAP). 2014;54p.
- [7] Wei S, Chen LQ, Taniyasu S, So MK, Murphy MB, Yamashita N, Yeung LWY, Lam PKS. Distribution of perfluorinated compounds in surface seawaters between Asia and Antarctica. *Mar Pollut Bull* 2007;54:1813–38.
- [8] United Nations Environment Program Secretariat of the Stockholm Convention on Persistent Organic Pollutants. The Nine New POPs: an introduction to the nine chemicals added to the Stockholm Convention by the Conference of the Parties at its fourth meeting. 2010. Available from: <http://chm.pops.int/Programmes/NewPOPs/Publications/tabid/695/language/en-US/Default.aspx> [Accessed: 2015-01-29]
- [9] Schiavone A, Corsolini S, Kannan K, Tao L, Trivelpiece W, Torres D Jr, Focardi S. Perfluorinated contaminants in fur seal pups and penguin eggs from South Shetland, Antarctica. *Sci Total Environ* 2009;407:3899–904.
- [10] Cai M, Yang H, Xie Z, Zhao Z, Wang F, Lu Z, Sturm R, Ebinghaus R. Per- and polyfluoroalkyl substances in snow, lake, surface runoff water and coastal seawater in Fildes Peninsula, King George Island, Antarctica. *J Hazard Mater* 2012;209–10:335–42.
- [11] Del Vento S, Halsall C, Gloia R, Jones K, Dachs J. Volatile per- and polyfluoroalkyl compounds in the remote atmosphere of the western Antarctic Peninsula: an indirect source of perfluoroalkyl acids to Antarctic waters? *Atmos Pollut Res* 2012;3:450–5.
- [12] Llorca M, Farre M, Tavano MS, Alonso B, Koremblit G, Barcelo D. Fate of a broad spectrum of perfluorinated compounds in soils and biota from Tierra del Fuego and Antarctica. *Environ Pollut* 2012;163:158–66.
- [13] Yamashita N, Kannan K, Taniyasu S, Horii Y, Petrick G, Gamo T. A global survey of perfluorinated acids in oceans. *Mar Pollut Bull* 2005;51:658–68.

- [14] Prevedouros K, Cousins IT, Buck RC, Korzeniowski SH. Sources, fate, and transport of perfluorocarboxylates. *Environ Sci Technol* 2006;40:32–44.
- [15] Zhao Z, Xie Z, Moller A, Sturm R, Tang J, Zhang G, Ebinghaus R. Distribution and long-range transport of polyfluoroalkyl substances in the Arctic, Atlantic Ocean and Antarctic coast. *Environ Pollut* 2012;170:71–7.
- [16] U.S. EPA: Fluorochemical Use, Distribution and Release Overview; Public Docket AR226-0550; 26 May 1999; St. Paul, MN: 3M Company. 1999.
- [17] Begley TH, White K, Honigfort P, Twaroski ML, Neches R, Walker RA. Perfluorochemicals: potential sources of and migration from food packaging. *Food Addit Contam* 2005;22:1023–31.
- [18] Lau C, Anitole K, Hodes C, Lai D, Pfahles-Hutchens A, Seed J. Perfluoroalkyl acids: a review of monitoring and toxicological findings. *Toxicol Sci* 2007;99:366–94.
- [19] Kelly BC, Ikonomou MG, Blair JD, Surridge B, Hoover D, Haviland L, Grace R, Gobas FAPC. Perfluoroalkyl contaminants in an Arctic marine food web: trophic magnification and wildlife exposure. *Environ Sci Technol* 2009;43:4037–43.
- [20] Armitage JM, Schenker U, Scheringer M, Martin JW, MacLeod M, Cousins IT. Modeling the global fate and transport of perfluorooctane sulfonate (PFOS) and precursor compounds in relation to temporal trends in wildlife exposure. *Environ Sci Technol* 2009;43:9274–80.
- [21] Giesy JP, Naile JE, Khim JS, Jones PD, Newsted JL. Aquatic toxicology of perfluorinated chemicals. In: Whitacre DM, ed. *Rev Environ Contam Toxicol, Rev Environ Contam Toxicol* 202. Springer Science + Business Media, LLC; 2010;p.1–52. DOI 10.1007/978-1-4419-1157-5_1.
- [22] Lim TC, Wang B, Huang J, Deng S, Yu G. Emission Inventory for PFOS in China: review of past methodologies and suggestions. *Sci World J* 2011;11:1963–80. DOI: 10.1100/2011/868156.
- [23] Carloni D. Perfluorooctane sulfonate (PFOS) production and use: past and current evidence. Vienna, Austria, 2009: United Nations Industrial Development Organization (UNIDO); 2009;56p. Available from: http://www.unido.org/fileadmin/user_media/Services/Environmental_Management/Stockholm_Convention/POPs/DC_Perfluorooctane%20Sulfonate%20Report.PDF [Accessed: 2015-01-29].
- [24] Wei Y. Obstacles in risk management of PFOS. In: *Informal Workshop on Stakeholders' Information Needs on Chemicals in Articles/Products*; 9–12 February 2009; Geneva, Switzerland: United Nations Environment Programme (UNEP); 2009. Available from: http://www.chem.unep.ch/unepsaicm/cheminprod_dec08/default.htm [Accessed: 2015-01-29].
- [25] Han W. PFOS related actions in China. *Workshop on Managing Perfluorinated Chemicals and Transitioning to Safer Alternatives*; 12–13 February 2009; Geneva,

- Switzerland: United Nations Environment Programme (UNEP); 2009. Available from: http://www.chem.unep.ch/unep/chemprod_dec08/PFCWorkshop/default.htm [Accessed: 2015-01-29].
- [26] Rattner BA, Scheuhammer AM, Elliott JE. History of wildlife toxicology and the interpretation of contaminant concentrations in tissues. In: Beyer WN, Meador JP, eds. *Environmental Contaminants in Biota: Interpreting Tissue Concentrations*. 2nd ed. Boca Raton, FL: CRC Press; 2011. p.9–44.
- [27] Calle P, Alvarado O, Monserrate L, Cevallos, JM, Calle NL, Alava JJ. Mercury accumulation in sediments and seabird feathers from the Antarctic Peninsula. *Mar Pollut Bull* 2015;91(2):410–7. DOI:10.1016/j.marpolbul.2014.10.009.
- [28] Meyer J, Jaspers VLB, Eens M, de Coen W. The relationship between perfluorinated chemical levels in the feathers and livers of birds from different trophic levels. *Sci Total Environ* 2009;407:5894–900. DOI:10.1016/j.scitotenv.2009.07.032.
- [29] Fujita T, Iwassa J, Hansch CA. New substituent constant, π , derived from partition coefficients. *J Am Chem Soc* 1964;86(23):5175–80.
- [30] Kah M, Brown CD. Log *D*: lipophilicity for ionisable compounds. *Chemosphere* 2008;72(10):1401–8.
- [31] Armitage JM, Arnot JA, Wania F, Mackay D. Development and evaluation of a mechanistic bioconcentration model for ionogenic organic chemicals in fish. *Environ Toxicol Chem* 2013;32(1):115–28.
- [32] Ostertag SK, Tague BA, Humphries MM, Tittlemier SA, Chan HM. Estimated dietary exposure to fluorinated compounds from traditional foods among Inuit in Nunavut, Canada. *Chemosphere* 2009;75:1165–72.
- [33] Arp HP, Niederer C, Goss KU. Predicting the partitioning behavior of various highly fluorinated compounds. *Environ Sci Technol* 2006;40:7298–304.
- [34] Bossi R, Riget FF, Dietz R. Temporal and spatial trends of perfluorinated compounds in ringed seal (*Phoca hispida*) from Greenland. *Environ Sci Technol* 2005;39:7416–22.
- [35] Dietz R, Bossi R, Riget FF, Sonne C, Born EW. Increasing perfluoroalkyl contaminants in east Greenland polar bears (*Ursus maritimus*): a new toxic threat to the Arctic bears. *Environ Sci Technol* 2008;42:2701–7.
- [36] Chen CL, Lu YL, Zhang X, Geng J, Wang TY, Shi YJ, Hu WY, Li J. A review of spatial and temporal assessment of PFOS and PFOA contamination in China. *Chem Ecol* 2009;25:163–77.
- [37] Reiner JL, O'Connell SG, Moors AJ, Kucklick JR, Becker PR, Keller JM. Spatial and temporal trends of perfluorinated compounds in Beluga Whales (*Delphinapterus leucas*) from Alaska. *Environ Sci Technol* 2011;45:8129–36.

- [38] Butt, CM, Muir DCG, Stirling I, Kwan M, Mabury SA. Rapid response of arctic ringed seals to changes in perfluoroalkyl production. *Environ Sci Technol* 2007;41:42–9.
- [39] Hart K, Gill VA, Kannan K. Temporal trends (1992–2007) of perfluorinated chemicals in northern sea otters (*Enhydra lutris kenyoni*) from South-Central Alaska. *Arch Environ Contam Toxicol* 2009;56:607–14.
- [40] Gobas FAPC, Morrison HA. Bioconcentration and bioaccumulation in the aquatic environment. In: Boethling R, Mackay D, eds. *Handbook of Property Estimation methods for chemicals: Environmental and Health Sciences*. Boca Raton, FL, USA: CRC Press LLC; 2000. p.189–231.
- [41] Gobas FAPC, de Wolf W, Verbruggen E, Plotzke K, Burkhard L. Revisiting bioaccumulation criteria for POPs and PBT assessments. *Integrat Environ Ass Manage* 2009;5(4):624–37.
- [42] Newsted JL, Jones PD, Coady KK, Giesy JP. Avian toxicity reference values for perfluorooctane sulfonate. *Environ Sci Technol* 2005;39:9357–62.
- [43] Wania F. A global mass balance analysis of the source of perfluorocarboxylic acids in the Arctic Ocean. *Environ Sci Technol* 2007;41(13):4529–35.
- [44] Dreyer A, Weinberg I, Temme C, Ebinghaus R. Polyfluorinated compounds in the atmosphere of the Atlantic and Southern Oceans: evidence for a global distribution. *Environ Sci Technol* 2009;43:6507–14.
- [45] Bengtson Nash S, Rintoul SR, Kawaguchi S, Staniland I, van den Hoff J, Tierney M, Bossi R. Perfluorinated compounds in the Antarctic region: ocean circulation provides prolonged protection from distant sources. *Environ Pollut* 2010;158(9):2985–91.
- [46] Ellis DA, Martin J, De Silva AO, Mabury SA, Hurley MD, Sulbaek Andersen MP, Wallington TJ. Degradation of fluorotelomer alcohols: a likely atmospheric source of perfluorinated carboxylic acids. *Environ Sci Technol* 2004;38(12):3316–21.
- [47] Wallington TJ, Hurley MD, Xia J, Wuebbles DJ, Sillmann S, Ito A, Penner JE, Ellis DA, Martin J, Mabury SA, Nielson OJ, Andersen MPS. Formation of C₇F₁₅COOH (PFOA) and other perfluorocarboxylic acids during the atmospheric oxidation of 8:2 fluorotelomer alcohol. *Environ Sci Technol* 2006;40(3):924–30.
- [48] Schenker U, Scheringer M, MacLeod M, Martin JW, Cousins IT, Hungerbühler K. Contribution of volatile precursor substances to the flux of perfluorooctanoate to the Arctic. *Environ Sci Technol* 2008;42(10):3710–6.
- [49] Armitage JM, Cousins IT, Buck RC, Prevedouros K, Russell MH, MacLeod M, Korzeniowski SH. Modeling global-scale fate and transport of perfluorooctanoate emitted from direct sources. *Environ Sci Technol* 2006;40(22):864–72.

- [50] Paul AG, Jones KC, Sweetman AJ. A first global production, emission, and environmental inventory for perfluorooctane sulfonate. *Environ Sci Technol* 2009;43:386–92.
- [51] Lee H, D'eon J, Mabury SA. Biodegradation of polyfluoroalkyl phosphates as a source of perfluorinated acids to the environment. *Environ Sci Technol* 2010;44:3305–10.
- [52] Benskin JP, Muir CG, Scott BF, Spencer C, De Silva AO, Kylin H, Martin JW, Morris A, Lohmann R, Tomy G, Rosenberg B, Taniyasu S, Yamashita N. Perfluoroalkyl acids in the Atlantic and Canadian Arctic Oceans. *Environ Sci Technol* 2012;46(11):5815–23.
- [53] Tomy GT, Tittlemier SA, Palace VP, Budakowski WR, Braekevelt E, Brinkworth L, Friesen K. Biotransformation of N-ethyl perfluorooctanesulfonamide by rainbow trout (*Onchorhynchus mykiss*) liver microsomes. *Environ Sci Technol* 2004;38:758–2.
- [54] Martin JW, Ellis DA, Mabury SA, Hurley MD, Wallington TJ. Atmospheric chemistry of perfluoroalkanesulfonamides: kinetic and product studies of the OH radical and Cl atom initiated oxidation of N-ethyl perfluorobutanesulfonamide. *Environ Sci Technol* 2006;40:864–72.
- [55] United Nations Environment Programme: Risk Management evaluation on perfluorooctane sulfonate. Report of the Persistent Organic Pollutants Review Committee on the work of its third meeting; 19–23 November, 2007. Geneva, Switzerland: United Nations Environment Programme (UNEP). 2007.
- [56] INAE (Instituto Antártico Ecuatoriano): Acerca del INAE [Internet]. 2015. Available from: <http://www.inae.gob.ec/index.php/sobre-inae> [Accessed: 2015-02-02].
- [57] COMINAP (Council of Managers of National Antarctic Programs): Instituto Antártico Ecuatoriano (INAE): About Ecuador's National Antarctic Program [Internet]. 2015. Available from: <https://www.comnap.aq/Members/ecuador/SitePages/Home.aspx> [Accessed: 2015-02-02].
- [58] [58]Hemmings AD. 'Environmental Management' as diplomatic method: the advancement of strategic national interest in Antarctica. In: Liggett D, Hemmings AD, eds. *Exploring Antarctic Values. Proceedings of the the workshop Exploring Linkages between Environmental Management and Value Systems: The Case of Antarctica*; 5 December 2011; Christchurch: University of Canterbury, Gateway Antarctica Special Publication Series 1301; 2013. p.70–89. Available from:<http://antarctica-ssag.org/wp-content/uploads/2013/05/SSAG-proceedings-2013.pdf>. [Accessed: 2015-02-02].
- [59] Instituto Antártico Ecuatoriano: Estudio de Impacto Ambiental, Ex-Post. Estación Científica Ecuatoriana Pedro Vicente Maldonado; Verano Austral, Enero-Marzo 2010 y 2011. Isla Greenwich, Shetland del Sur, Antártida: Soluciones Ambientales Totales (SAMBITO). 2011. p. 117. Available from: <http://es.slideshare.net/sambitoeco/eia-ex-post-pevima> [Accessed: 2015-02-02].

- [60] Tanabe S, Hidaka H, Tatsukawa, R. PCBs and chlorinated hydrocarbon pesticides in Antarctic atmosphere and hydrosphere. *Chemosphere* 1983;12:277–88.
- [61] Larsson P, Järnmark C, Södergren A. PCBs and chlorinated pesticides in the atmosphere and aquatic organisms of Ross Island, Antarctica. *Marine Pollut Bull* 1992;25:281–7.
- [62] Sen Gupta R, Sarkar A, Kureishey TW. PCBs and organochlorine pesticides in krill, birds and water from Antarctica. *Deep Sea Res Part II Top Stud Oceanog* 1996;43:119–26.
- [63] Alava JJ, Ikonomou MG, Riofrío Briceño M, Gobas FAPC. Estudio de contaminantes orgánicos persistentes (COPs) en la Antártica: Contaminantes Perfluoroalquilos (Perfluoroalkyl Contaminants, PFCs). Informe Final 2009. Ministerio de Defensa Nacional & Instituto Antártico Ecuatoriano (INAE); 15 Septiembre 2009. Guayaquil: INAE. 2009. p. 20.

